## SYNTHESIS OF PYRYLIUM SALTS

FROM TRISUBSTITUTED 1,4-PENTADIYN-3-OLS WITH HClo<sup>1)</sup>

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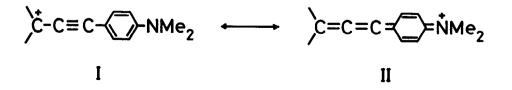
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Summary: The synthesis of 2,4,6-trisubstituted pyrylium salts from 1,3,5-trisubstituted 1,4-pentadiyn-3-ols with HClO<sub>4</sub> is reported for the first time.

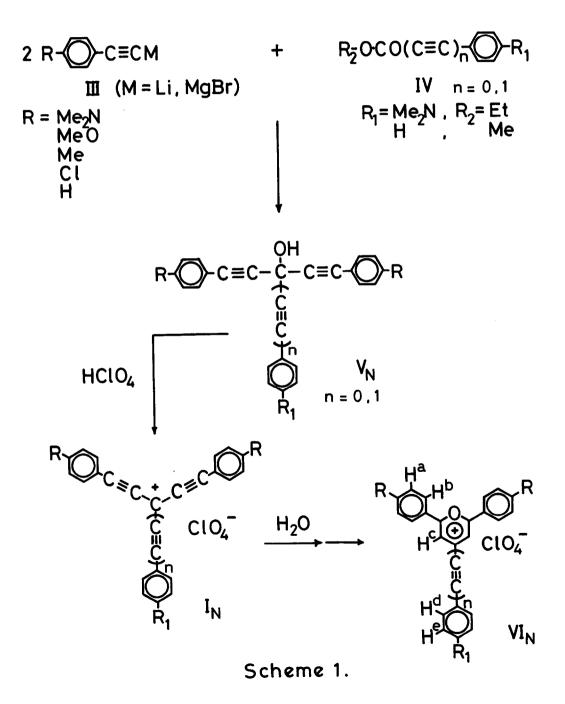
Throughout the study on the monoethynologs<sup>1,2)</sup> of Malachite Green, Crystal Violet, and their related compounds, the synthesis of bis- and trisethynologs ( $I_N$ ) of the above dyes has been the focus of considerable interest. These ethynologs may be expected to form a new resonance system (I  $\leftrightarrow$  II) containing an allene-quinoid structure (II) realized by delocalization of a positive charge. Then, the alcohols ( $V_N$ ) were prepared for the precursors of  $I_N$  and treated with HClo<sub>4</sub>. However, it was found that the pyrylium salts (VI<sub>N</sub>) were obtained in moderate yields instead of the anticipated ethynologs ( $I_N$ ) (Scheme 1).

In this communication, we wish to report a novel synthesis of 2,4,6-trisubstituted pyrylium salts (VI<sub>N</sub>) from 1,3,5-trisubstituted 1,4-pentadiyn-3-ols (V<sub>N</sub>) by trearment with  $HClO_4$ .

Reaction of the *p*-substituted phenylacetylides (III)<sup>3)</sup> (2 mol) with the corresponding esters (IV) [*e.g.*, methyl *p*-dimethylaminobenzoate or ethyl *p*-dimethylaminophenylpropiolate<sup>3)</sup>] (1 mol) in THF at -78 °C gave  $V_N$  as depicted in Scheme 1.<sup>4</sup>)



Treatment of  $V_N$  with  $HClo_4$  (70 wt%) in benzene - EtOH (2 : 1, vol) yielded  $VI_N$  (Table 1). In Table 1,  $VI_3$ ,  $VI_4$ , and  $VI_7$  are new compounds, whose structures were clearly assigned from the data obtained.<sup>4,5)</sup> The other pyrylium salts were identified distinctly by comparing their physical data with those of the literature or/and the authentic samples prepared according to the conventional methods.<sup>6,7)</sup>



N	n	R	Rl	v <sub>N</sub>	VI <sub>N</sub>	mp (°C)	yield (%)
1	0	Me <sub>2</sub> N	Н	v <sub>1</sub>	VI1	264 - 266 <sup>a)</sup>	67
2	0	MeO	н	v <sub>2</sub>	VI <sub>2</sub>	ca. 275 <sup>b)</sup>	89
3	0	Me	н	v <sub>3</sub>	vı <sub>3</sub>	278 – 279	79
4	0	Cl	Н	v <sub>4</sub>	VI4	293 - 295	52
5	0	н	н	v <sub>5</sub>	VI5	289 – 292 <sup>C)</sup>	32 <sup>d)</sup>
6	1	н	н	v <sub>6</sub>	VI <sub>6</sub>	179–181 <sup>e)</sup>	68
7	1	$Me_2^N$	$^{Me}2^{N}$	v <sub>7</sub>	VI <sub>7</sub>	ca. 230	59

Table 1. Synthesis of Pyrylium Salts (VI<sub>N</sub>) from Alcohols (V<sub>N</sub>)

a) Reference 7): 264-265 °C.

b) Reference 7): 274-275 °C.

c) J. A. VanAllen and G. A. Reynolds, J. Org. Chem., <u>33</u>, 1102 (1968): 293-295 °C.

d) Based on crude  $V_5$ .

e) J. J. Doney, C. H. Chen, and H. R. Luss, Tetrahedron Lett., <u>23</u>, 1747 (1982): 179-181 °C.

The formation of the pyrylium salts  $(VI_N)$  may be presumed as follows: In the acidic medium the alcohols  $(V_N)$  give the labile cations  $(I_N)$ , <sup>8)</sup> which undergo rapidly Meyer-Schuster rearrangement<sup>9)</sup> and then yield a series of compounds, *e.g.*, 2-pentene-1,5-dione, 2-penten-4-yn-1-one derivatives or/and their enols, by an addition of water.<sup>10)</sup> Subsequently, a pyrylium ring is yielded as a result of intramolecular condensation and succeeding aromatization reactions.

Interestingly, the effect of the substituents (R) was observed in this pyrylium ring formation reaction. The alcohols ( $V_3$ ,  $V_4$ , and  $V_7$ ) bearing strong electron donating groups cyclized quite easily to the corresponding pyrylium salts at room temperature. However, the others needed heating for several hours to complete the above reactions. From these results, the order of the ease of the cyclization was as given follows: Me<sub>2</sub>N > MeO > Me > H  $\approx$  Cl

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## References and Notes

- Ethynologs of Triphenylmethane Dyes. Part II. For Part I, see S. Akiyama, K. Yoshida, M. Hayashida, K. Nakashima, S. Nakatsuji, and M. Iyoda, Chem. Lett., <u>1981</u>, 311.
- S. Akiyama, K. Nakashima, S. Nakatsuji, M. Hamada, and Y. Izaki, Bull. Chem. Soc. Jpn., <u>56</u>, 947 (1983).
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- 4) The IR, <sup>1</sup>H NMR, and mass spectral data of all crystalline compounds described in this communication are consistent with the assigned structures. Acceptable combustion data were also obtained.
- 5) Spectral data of the new compounds  $(VI_3, VI_4, and VI_7)$ :  $VI_3$ , red crystals, IR (Nujol) 1615 cm<sup>-1</sup>; UV (CH<sub>2</sub>Cl<sub>2</sub>) 373 ( $\varepsilon$  40700), 445 (30100), and 532 nm (3580); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  2.55 (6H, s, Me), 7.21-7.46 and 7.74-8.00 (3H, m, H<sup>e</sup> and R<sub>1</sub>=H), 7.62 (4H, d, J = 9 Hz, H<sup>a</sup>), 8.46-8.57 (2H, m, H<sup>d</sup>), 8.52 (4H, d, J = 9 Hz, H<sup>b</sup>), and 9.06 (2H, s, H<sup>C</sup>); VI<sub>4</sub>, yellow crystals, IR (Nujol) 1620 cm<sup>-1</sup>; UV (CH<sub>2</sub>Cl<sub>2</sub>) 303 ( $\varepsilon$  7470), 377 (14100), and 432 nm (9330); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  7.71-7.88 and 8.52-8.63 (13H, m, aromatic protons) and 9.13 (2H, s, H<sup>C</sup>); VI<sub>7</sub>, deep violet crystals, IR (KBr) 2170 and 1570 cm<sup>-1</sup>; UV (CH<sub>2</sub>Cl<sub>2</sub>) 590 ( $\varepsilon$  76600) and 640 nm (63000); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  3.07 (6H, s, Me<sub>2</sub>N), 3.15 (12H, s, Me<sub>2</sub>N), 6.84 (2H, d, J = 9 Hz, H<sup>e</sup>), 6.92 (4H, d, J = 9.5 Hz, H<sup>a</sup>), 7.55 (2H, d, J = 9 Hz, H<sup>d</sup>), 8.00 (2H, s, H<sup>C</sup>), and 8.11 (4H, d, J = 9.5 Hz, H<sup>b</sup>).
- 6) For a review of pyrylium salts, see A. T. Balaban, G. W. Fischer, A. Dinculescu, A. V. Koblik, G. N. Dorofeenko, V.V. Mezheritskii, and W. Schroth, "Pyrylium Salts: Syntheses, Reactions, and Physical Properties," in "Avdances in Heterocyclic Chemistry, Supplement 2," ed by A. R. Katritzky, Academic Press, New York (1982).
- E.g., for VI<sub>6</sub>, cf. R. Wizinger, S. Losinger, and P. Ulrich, Helv. Chim. Acta, 39, 5 (1956).
- 8) Attempts to characterize the cations  $(I_N)$  by the <sup>1</sup>H NMR measurements of  $V_N$  in various anhydrous acidic media  $(CF_3CO_2D, CD_3CO_2D, and FSO_3H-SbF_5)$  at -30 -70 °C have been so far impossible.
- 9) H. Krauch and W. Kunz, "Organic Name Reactions," John Wiley & Sons, Inc., New York (1964) p. 313, and references cited therein.
- 10) In the study on the monoethynologs of triphenylmethane dyes,<sup>1)</sup> we have isolated the ketone, l,l-diphenyl-2-(p-dimethylaminobenzoyl)ethylene, which was obtained by the treatment of 3-(p-dimethylaminophenyl)-1,l-diphenyl-2propyn-1-ol with CF<sub>3</sub>CO<sub>2</sub>H, followed by neutralization with aqueous Na<sub>2</sub>CO<sub>3</sub>. (Received in Japan 16 July 1984)